

Bis(pyridine-2-carbaldehyde thiosemi-carbazone)zinc(II) dinitrate dihydrate

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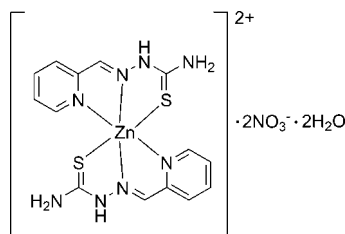
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 Key indicators: single-crystal X-ray study; $T = 120$ K; mean $\sigma(\text{C}-\text{C}) = 0.008$ Å; R factor = 0.057; wR factor = 0.123; data-to-parameter ratio = 13.8.

The asymmetric unit of the title compound, $[\text{Zn}(\text{C}_7\text{H}_8\text{N}_4\text{S})_2](\text{NO}_3)_2 \cdot 2\text{H}_2\text{O}$, contains two $\text{Zn}(\text{pht})_2$ cations (pht is pyridine-2-carbaldehyde thiosemicarbazone), four nitrate anions and four water molecules. In the cations, each Zn^{II} ion adopts a distorted octahedral coordination geometry, being chelated by two tridentate pht ligands. In the crystal, the cations, anions and water molecules are connected *via* $\text{O}-\text{H} \cdots \text{O}$ and $\text{N}-\text{H} \cdots \text{O}$ hydrogen bonds into a three-dimensional network.

Related literature

For related structures, see: Antholine *et al.* (1977); Ainscough *et al.* (1998).



Experimental

Crystal data

 $[\text{Zn}(\text{C}_7\text{H}_8\text{N}_4\text{S})_2](\text{NO}_3)_2 \cdot 2\text{H}_2\text{O}$
 $M_r = 585.89$

 Monoclinic, $P2_1/c$
 $a = 21.4623$ (14) Å

 $b = 16.6324$ (12) Å

 $c = 13.2764$ (10) Å

 $\beta = 102.876$ (2)°

 $V = 4620.1$ (6) Å³
 $Z = 8$

 Mo $K\alpha$ radiation

 $\mu = 1.31$ mm⁻¹
 $T = 120$ K

 $0.25 \times 0.22 \times 0.20$ mm

Data collection

Bruker SMART APEX CCD diffractometer

Absorption correction: multi-scan (SADABS; Sheldrick, 2008a)

 $T_{\text{min}} = 0.736$, $T_{\text{max}} = 0.780$

22904 measured reflections

8732 independent reflections

 5113 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.077$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.057$
 $wR(F^2) = 0.123$
 $S = 1.02$

8732 reflections

631 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.81$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.43$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H} \cdots A$ | $D-\text{H}$ | $\text{H} \cdots A$ | $D \cdots A$ | $D-\text{H} \cdots A$ |
|-------------------------------|--------------|---------------------|--------------|-----------------------|
| N3—H3B···O7 | 0.86 | 2.19 | 2.966 (6) | 150 |
| N4—H4B···O7 | 0.86 | 2.17 | 2.957 (6) | 153 |
| N4—H4C···O3W | 0.86 | 2.03 | 2.840 (5) | 157 |
| N7—H7A···O1 | 0.86 | 1.96 | 2.813 (6) | 172 |
| N8—H8A···O2 | 0.86 | 2.08 | 2.906 (6) | 160 |
| N8—H8B···O6 ⁱ | 0.86 | 2.04 | 2.870 (6) | 163 |
| N11—H11B···O10 | 0.86 | 1.87 | 2.710 (5) | 167 |
| N12—H12B···O11 | 0.86 | 2.02 | 2.876 (6) | 173 |
| N12—H12C···O8 ⁱⁱ | 0.86 | 2.27 | 3.114 (6) | 168 |
| N15—H15A···O4W ⁱⁱⁱ | 0.86 | 1.87 | 2.733 (5) | 178 |
| N16—H16A···O6 ⁱⁱⁱ | 0.86 | 2.11 | 2.909 (6) | 155 |
| N16—H16B···O4 ^{iv} | 0.86 | 2.02 | 2.873 (5) | 173 |
| O1W—H1WA···O4 ^v | 0.85 | 2.32 | 3.042 (5) | 144 |
| O1W—H1WB···O3 | 0.85 | 2.39 | 3.041 (6) | 134 |
| O2W—H2WA···O5 ^{iv} | 0.85 | 2.33 | 2.927 (6) | 128 |
| O2W—H2WB···O3 | 0.85 | 2.47 | 3.141 (6) | 136 |
| O3W—H3WA···O10 ^{vi} | 0.85 | 2.22 | 2.800 (5) | 126 |
| O3W—H3WB···O9 ^{vii} | 0.85 | 2.24 | 2.978 (6) | 145 |
| O4W—H4WA···O2W | 0.85 | 2.09 | 2.746 (6) | 133 |
| O4W—H4WB···O6 | 0.85 | 2.55 | 2.958 (5) | 111 |

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $x, -y + \frac{1}{2}, z + \frac{1}{2}$; (iii) $x, -y + \frac{3}{2}, z + \frac{1}{2}$; (iv) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$; (v) $x, -y + \frac{3}{2}, z - \frac{1}{2}$; (vi) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$; (vii) $-x, y + \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008b); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008b); molecular graphics: ORTEPIII (Burnett & Johnson, 1996) and PLATON (Spek, 2009); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5358).

References

- Ainscough, E. W., Brodie, A. M., Denny, W. A., Finlay, G. J. & Ranford, J. D. (1998). *J. Inorg. Biochem.* **70**, 175–185.
- Antholine, W. E., Knight, J. M. & Petering, D. H. (1977). *Inorg. Chem.* **16**, 569–574.
- Bruker (2000). SMART and SAINT-Plus. Bruker AXS Inc., Madison, Wisconsin, USA.
- Burnett, M. N. & Johnson, C. K. (1996). ORTEPIII. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Sheldrick, G. M. (2008a). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (2008b). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

supplementary materials

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Bis(pyridine-2-carbaldehyde thiosemicarbazone)zinc(II) dinitrate dihydrate

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Comment

The examination of the antitumor properties of N-heterocyclic carboxaldehyde thiosemicarbazones has been extended to the consideration of some of their first-row transition metal complexes (Antholine *et al.* 1977, Ainscough *et al.* 1998). Meanwhile, zinc plays an important role in various biological systems and is a vital component an essential cofactor, critical for numerous cellular processes and may be a major regulatory ion in the metabolism of cells. Herein, we report a Zn^{II} complex, Zn(pht)₂·2NO₃·2H₂O (pht= 2-(pyridine-2-carbaldehyde)hydrazinecarbothioamide).

The asymmetric unit of the title compound, contains two Zn(pht)₂ cations (pht = 2-(pyridine-2-carbaldehyde)hydrazinecarbothioamide), which have the similar structure, four nitrates and four free water molecules. In the Zn(pht)₂ cations, each Zn^{II} ions adopts a distorted octahedral coordination geometry, being chelated by two tridentate pht ligands. In a pht, all the atoms are approximatively coplanar, and the distances of the C=N bond are 1.269 (6)–1.283 (6) Å, which are shorter than those of C—N bond (1.303 (7)–1.369 (6) Å), being considered to have full double-bond character. In packing, All the Zn(pht)₂ cations, nitrates and water molecules are linked each other *via* O—H···O and N—H···O hydrogen bonds into a three-dimensional supramolecular network.

Experimental

A mixture of pht (0.036 g, 0.2 mmol), Zn(NO₃)₂·6H₂O (0.029 g, 0.1 mmol), and water (8 ml) were heated in a 15 ml Teflon-lined vessel at 120 ° for 3 days, followed by slow cooling (5 ° h⁻¹) to room temperature. After filtration, colorless block crystals were collected and dried in air (0.018 g, yield *ca* 30.7% based on pht).

Refinement

All H atoms attached to C and N atoms were fixed geometrically and treated as riding with C—H = 0.93 Å and N—H = 0.86 Å with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C or N})$. H atoms of water molecules were located in difference Fourier maps and included in the subsequent refinement with O—H = 0.85 Å with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{O})$.

Figures

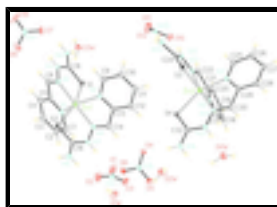


Fig. 1. View of the asymmetric unit of the title compound with the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms are represented as small spheres of arbitrary radii.

Bis(pyridine-2-carbaldehyde thiosemicarbazone)zinc(II) dinitrate dihydrate

Crystal data

| | |
|--|---|
| $[\text{Zn}(\text{C}_7\text{H}_8\text{N}_4\text{S})_2](\text{NO}_3)_2 \cdot 2\text{H}_2\text{O}$ | $F(000) = 2400$ |
| $M_r = 585.89$ | $D_x = 1.685 \text{ Mg m}^{-3}$ |
| Monoclinic, $P2_1/c$ | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| Hall symbol: -P 2ybc | Cell parameters from 785 reflections |
| $a = 21.4623 (14) \text{ \AA}$ | $\theta = 2.4\text{--}28.0^\circ$ |
| $b = 16.6324 (12) \text{ \AA}$ | $\mu = 1.31 \text{ mm}^{-1}$ |
| $c = 13.2764 (10) \text{ \AA}$ | $T = 120 \text{ K}$ |
| $\beta = 102.876 (2)^\circ$ | Block, colorless |
| $V = 4620.1 (6) \text{ \AA}^3$ | $0.25 \times 0.22 \times 0.20 \text{ mm}$ |
| $Z = 8$ | |

Data collection

| | |
|--|--|
| Bruker SMART APEX CCD diffractometer | 8732 independent reflections |
| Radiation source: fine-focus sealed tube graphite | 5113 reflections with $I > 2\sigma(I)$ |
| φ and ω scan | $R_{\text{int}} = 0.077$ |
| Absorption correction: multi-scan (SADABS; Sheldrick, 2008a) | $\theta_{\text{max}} = 26.0^\circ$, $\theta_{\text{min}} = 1.6^\circ$ |
| $T_{\text{min}} = 0.736$, $T_{\text{max}} = 0.780$ | $h = -25 \rightarrow 21$ |
| 22904 measured reflections | $k = -20 \rightarrow 12$ |
| | $l = -16 \rightarrow 16$ |

Refinement

| | |
|---------------------------------|--|
| Refinement on F^2 | Primary atom site location: structure-invariant direct methods |
| Least-squares matrix: full | Secondary atom site location: difference Fourier map |
| $R[F^2 > 2\sigma(F^2)] = 0.057$ | Hydrogen site location: inferred from neighbouring sites |
| $wR(F^2) = 0.123$ | H-atom parameters constrained |
| $S = 1.02$ | $w = 1/[\sigma^2(F_o^2) + (0.0354P)^2]$ |
| 8732 reflections | where $P = (F_o^2 + 2F_c^2)/3$ |
| 631 parameters | $(\Delta/\sigma)_{\text{max}} = 0.029$ |
| 0 restraints | $\Delta\rho_{\text{max}} = 0.81 \text{ e \AA}^{-3}$ |
| | $\Delta\rho_{\text{min}} = -0.43 \text{ e \AA}^{-3}$ |

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations

between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|-------------|-------------|--------------|----------------------------------|
| Zn1 | 0.25244 (3) | 0.23919 (3) | 0.57658 (4) | 0.02381 (17) |
| Zn2 | 0.22452 (3) | 0.75501 (3) | 0.82766 (4) | 0.02208 (17) |
| S1 | 0.18156 (7) | 0.30117 (8) | 0.42437 (10) | 0.0301 (4) |
| S2 | 0.32904 (7) | 0.17217 (8) | 0.48824 (11) | 0.0330 (4) |
| S3 | 0.14317 (7) | 0.65650 (8) | 0.85347 (10) | 0.0280 (3) |
| S4 | 0.30520 (7) | 0.65526 (8) | 0.80976 (10) | 0.0280 (4) |
| C1 | 0.3251 (3) | 0.1529 (3) | 0.7846 (4) | 0.0339 (15) |
| H1A | 0.3582 | 0.1887 | 0.7831 | 0.041* |
| C2 | 0.3318 (3) | 0.0998 (3) | 0.8668 (4) | 0.0378 (16) |
| H2A | 0.3684 | 0.1004 | 0.9195 | 0.045* |
| C3 | 0.2829 (3) | 0.0460 (3) | 0.8692 (4) | 0.0386 (16) |
| H3A | 0.2863 | 0.0094 | 0.9231 | 0.046* |
| C4 | 0.2294 (3) | 0.0477 (3) | 0.7902 (4) | 0.0332 (15) |
| H4A | 0.1959 | 0.0123 | 0.7900 | 0.040* |
| C5 | 0.2258 (3) | 0.1028 (3) | 0.7106 (4) | 0.0266 (13) |
| C6 | 0.1702 (3) | 0.1088 (3) | 0.6264 (4) | 0.0277 (13) |
| H6A | 0.1362 | 0.0731 | 0.6202 | 0.033* |
| C8 | 0.1178 (2) | 0.2397 (3) | 0.4168 (4) | 0.0223 (12) |
| C9 | 0.1696 (3) | 0.3390 (3) | 0.7026 (4) | 0.0270 (13) |
| H9A | 0.1368 | 0.3033 | 0.6752 | 0.032* |
| C10 | 0.1577 (3) | 0.4020 (3) | 0.7660 (4) | 0.0285 (14) |
| H10A | 0.1173 | 0.4086 | 0.7797 | 0.034* |
| C11 | 0.2065 (3) | 0.4540 (3) | 0.8076 (4) | 0.0330 (15) |
| H11A | 0.1996 | 0.4961 | 0.8499 | 0.040* |
| C12 | 0.2655 (3) | 0.4431 (3) | 0.7859 (4) | 0.0298 (14) |
| H12A | 0.2992 | 0.4773 | 0.8137 | 0.036* |
| C13 | 0.2735 (3) | 0.3807 (3) | 0.7222 (3) | 0.0200 (12) |
| C14 | 0.3341 (3) | 0.3668 (3) | 0.6934 (4) | 0.0256 (13) |
| H14A | 0.3701 | 0.3973 | 0.7214 | 0.031* |
| C16 | 0.3920 (3) | 0.2337 (3) | 0.5309 (4) | 0.0288 (13) |
| C17 | 0.3197 (3) | 0.8776 (3) | 0.7546 (4) | 0.0286 (14) |
| H17A | 0.3479 | 0.8639 | 0.8161 | 0.034* |
| C18 | 0.3399 (3) | 0.9322 (3) | 0.6897 (4) | 0.0300 (14) |
| H18A | 0.3812 | 0.9530 | 0.7059 | 0.036* |
| C19 | 0.2978 (3) | 0.9545 (3) | 0.6017 (4) | 0.0334 (15) |
| H19A | 0.3097 | 0.9922 | 0.5577 | 0.040* |
| C20 | 0.2369 (3) | 0.9208 (3) | 0.5776 (4) | 0.0291 (14) |
| H20A | 0.2073 | 0.9359 | 0.5182 | 0.035* |

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|------|--------------|------------|------------|-------------|
| C21 | 0.2214 (2) | 0.8643 (3) | 0.6445 (4) | 0.0207 (12) |
| C22 | 0.1609 (3) | 0.8202 (3) | 0.6203 (4) | 0.0256 (13) |
| H22A | 0.1292 | 0.8320 | 0.5621 | 0.031* |
| C24 | 0.0965 (3) | 0.6578 (3) | 0.7329 (4) | 0.0232 (13) |
| C25 | 0.1337 (3) | 0.8917 (3) | 0.8776 (4) | 0.0278 (13) |
| H25A | 0.1075 | 0.8782 | 0.8141 | 0.033* |
| C26 | 0.1147 (3) | 0.9538 (3) | 0.9340 (4) | 0.0285 (14) |
| H26A | 0.0771 | 0.9819 | 0.9077 | 0.034* |
| C27 | 0.1520 (3) | 0.9729 (3) | 1.0289 (4) | 0.0320 (14) |
| H27A | 0.1398 | 1.0135 | 1.0686 | 0.038* |
| C28 | 0.2086 (3) | 0.9305 (3) | 1.0648 (4) | 0.0320 (14) |
| H28A | 0.2350 | 0.9427 | 1.1286 | 0.038* |
| C29 | 0.2246 (3) | 0.8702 (3) | 1.0038 (4) | 0.0235 (13) |
| C30 | 0.2841 (3) | 0.8244 (3) | 1.0344 (4) | 0.0258 (13) |
| H30A | 0.3132 | 0.8343 | 1.0963 | 0.031* |
| C32 | 0.3575 (3) | 0.6699 (3) | 0.9239 (4) | 0.0242 (13) |
| N1 | 0.2738 (2) | 0.1546 (2) | 0.7083 (3) | 0.0270 (11) |
| N2 | 0.1698 (2) | 0.1650 (2) | 0.5611 (3) | 0.0234 (10) |
| N3 | 0.1180 (2) | 0.1775 (2) | 0.4824 (3) | 0.0273 (11) |
| H3B | 0.0855 | 0.1460 | 0.4745 | 0.033* |
| N4 | 0.06559 (19) | 0.2474 (2) | 0.3439 (3) | 0.0275 (11) |
| H4B | 0.0346 | 0.2140 | 0.3405 | 0.033* |
| H4C | 0.0624 | 0.2858 | 0.2997 | 0.033* |
| N5 | 0.2263 (2) | 0.3291 (2) | 0.6810 (3) | 0.0225 (10) |
| N6 | 0.3358 (2) | 0.3111 (2) | 0.6284 (3) | 0.0243 (11) |
| N7 | 0.3911 (2) | 0.2952 (2) | 0.5973 (3) | 0.0287 (11) |
| H7A | 0.4247 | 0.3239 | 0.6195 | 0.034* |
| N8 | 0.4457 (2) | 0.2266 (3) | 0.4975 (4) | 0.0427 (13) |
| H8A | 0.4766 | 0.2598 | 0.5183 | 0.051* |
| H8B | 0.4495 | 0.1888 | 0.4552 | 0.051* |
| N9 | 0.2622 (2) | 0.8438 (2) | 0.7334 (3) | 0.0231 (10) |
| N10 | 0.15389 (19) | 0.7643 (2) | 0.6836 (3) | 0.0195 (10) |
| N11 | 0.1020 (2) | 0.7152 (2) | 0.6629 (3) | 0.0244 (11) |
| H11B | 0.0732 | 0.7204 | 0.6067 | 0.029* |
| N12 | 0.0521 (2) | 0.6028 (2) | 0.7020 (3) | 0.0316 (12) |
| H12B | 0.0286 | 0.6047 | 0.6404 | 0.038* |
| H12C | 0.0466 | 0.5650 | 0.7433 | 0.038* |
| N13 | 0.1877 (2) | 0.8507 (2) | 0.9102 (3) | 0.0229 (10) |
| N14 | 0.2938 (2) | 0.7703 (2) | 0.9722 (3) | 0.0212 (10) |
| N15 | 0.3479 (2) | 0.7262 (2) | 0.9943 (3) | 0.0256 (11) |
| H15A | 0.3757 | 0.7333 | 1.0511 | 0.031* |
| N16 | 0.4098 (2) | 0.6278 (2) | 0.9496 (3) | 0.0328 (12) |
| H16A | 0.4360 | 0.6362 | 1.0079 | 0.039* |
| H16B | 0.4182 | 0.5916 | 0.9083 | 0.039* |
| N17 | 0.5277 (3) | 0.4141 (3) | 0.5916 (4) | 0.0421 (13) |
| N18 | 0.5183 (2) | 0.9676 (3) | 0.6417 (3) | 0.0291 (11) |
| N19 | -0.0127 (3) | 0.0485 (3) | 0.3698 (4) | 0.0358 (12) |
| N20 | -0.0031 (2) | 0.6531 (3) | 0.4316 (4) | 0.0358 (12) |
| O1 | 0.4955 (2) | 0.4005 (3) | 0.6560 (3) | 0.0628 (14) |

| | | | | |
|------|---------------|------------|------------|-------------|
| O2 | 0.5329 (2) | 0.3626 (3) | 0.5256 (4) | 0.0677 (15) |
| O3 | 0.5557 (2) | 0.4799 (3) | 0.5909 (3) | 0.0544 (13) |
| O4 | 0.57090 (17) | 0.9985 (2) | 0.6837 (3) | 0.0329 (10) |
| O5 | 0.46998 (18) | 1.0110 (2) | 0.6105 (3) | 0.0406 (11) |
| O6 | 0.51370 (17) | 0.8926 (2) | 0.6304 (3) | 0.0335 (10) |
| O7 | -0.01433 (19) | 0.1186 (2) | 0.4029 (3) | 0.0448 (11) |
| O8 | 0.0395 (2) | 0.0142 (3) | 0.3794 (3) | 0.0516 (12) |
| O9 | -0.0623 (2) | 0.0145 (3) | 0.3268 (3) | 0.0648 (15) |
| O10 | 0.02421 (18) | 0.7177 (2) | 0.4717 (3) | 0.0395 (11) |
| O11 | -0.01805 (19) | 0.6012 (2) | 0.4904 (3) | 0.0430 (11) |
| O12 | -0.01414 (18) | 0.6431 (2) | 0.3370 (3) | 0.0441 (11) |
| O1W | 0.62343 (19) | 0.4712 (3) | 0.4129 (3) | 0.0608 (13) |
| H1WA | 0.6112 | 0.4562 | 0.3504 | 0.073* |
| H1WB | 0.6057 | 0.4444 | 0.4533 | 0.073* |
| O2W | 0.52013 (19) | 0.6250 (3) | 0.7187 (3) | 0.0588 (13) |
| H2WA | 0.5064 | 0.6175 | 0.7733 | 0.071* |
| H2WB | 0.5496 | 0.5995 | 0.6990 | 0.071* |
| O3W | 0.04209 (17) | 0.3374 (2) | 0.1575 (3) | 0.0400 (10) |
| H3WA | 0.0152 | 0.3325 | 0.1001 | 0.048* |
| H3WB | 0.0337 | 0.3847 | 0.1746 | 0.048* |
| O4W | 0.43806 (18) | 0.7533 (2) | 0.6735 (3) | 0.0542 (12) |
| H4WA | 0.4679 | 0.7266 | 0.6564 | 0.065* |
| H4WB | 0.4545 | 0.7897 | 0.7161 | 0.065* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|------------|------------|-------------|------------|-------------|
| Zn1 | 0.0256 (4) | 0.0197 (3) | 0.0261 (3) | -0.0050 (3) | 0.0059 (3) | -0.0023 (3) |
| Zn2 | 0.0243 (4) | 0.0183 (3) | 0.0222 (3) | 0.0012 (3) | 0.0023 (3) | -0.0007 (3) |
| S1 | 0.0311 (10) | 0.0296 (8) | 0.0283 (7) | -0.0100 (7) | 0.0039 (7) | 0.0025 (6) |
| S2 | 0.0338 (10) | 0.0279 (8) | 0.0388 (8) | -0.0050 (7) | 0.0116 (7) | -0.0111 (6) |
| S3 | 0.0340 (10) | 0.0228 (7) | 0.0260 (7) | -0.0015 (7) | 0.0038 (6) | 0.0007 (6) |
| S4 | 0.0302 (9) | 0.0229 (7) | 0.0279 (7) | 0.0048 (7) | 0.0005 (6) | -0.0041 (6) |
| C1 | 0.040 (4) | 0.024 (3) | 0.034 (3) | -0.002 (3) | 0.001 (3) | -0.003 (3) |
| C2 | 0.059 (5) | 0.025 (3) | 0.028 (3) | 0.016 (3) | 0.006 (3) | -0.005 (3) |
| C3 | 0.065 (5) | 0.021 (3) | 0.036 (3) | 0.007 (3) | 0.025 (3) | 0.004 (3) |
| C4 | 0.045 (4) | 0.021 (3) | 0.038 (3) | 0.008 (3) | 0.019 (3) | 0.003 (3) |
| C5 | 0.035 (4) | 0.017 (3) | 0.034 (3) | 0.004 (3) | 0.019 (3) | 0.001 (2) |
| C6 | 0.029 (4) | 0.015 (3) | 0.043 (3) | -0.005 (3) | 0.015 (3) | -0.001 (3) |
| C8 | 0.023 (3) | 0.020 (3) | 0.025 (3) | 0.002 (3) | 0.008 (2) | -0.006 (2) |
| C9 | 0.038 (4) | 0.018 (3) | 0.025 (3) | -0.004 (3) | 0.008 (3) | 0.004 (2) |
| C10 | 0.032 (4) | 0.031 (3) | 0.026 (3) | 0.010 (3) | 0.012 (3) | 0.012 (2) |
| C11 | 0.055 (5) | 0.018 (3) | 0.025 (3) | 0.007 (3) | 0.005 (3) | 0.000 (2) |
| C12 | 0.041 (4) | 0.019 (3) | 0.028 (3) | -0.002 (3) | 0.004 (3) | -0.003 (2) |
| C13 | 0.023 (4) | 0.019 (3) | 0.017 (3) | 0.002 (3) | 0.004 (2) | 0.003 (2) |
| C14 | 0.030 (4) | 0.021 (3) | 0.023 (3) | -0.005 (3) | 0.001 (3) | 0.002 (2) |
| C16 | 0.029 (4) | 0.024 (3) | 0.034 (3) | 0.002 (3) | 0.009 (3) | -0.001 (2) |
| C17 | 0.027 (4) | 0.022 (3) | 0.035 (3) | 0.005 (3) | 0.004 (3) | -0.005 (3) |

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| C18 | 0.025 (4) | 0.023 (3) | 0.044 (3) | -0.003 (3) | 0.012 (3) | -0.005 (3) |
| C19 | 0.051 (4) | 0.021 (3) | 0.034 (3) | -0.013 (3) | 0.022 (3) | -0.006 (3) |
| C20 | 0.040 (4) | 0.026 (3) | 0.023 (3) | -0.005 (3) | 0.013 (3) | 0.000 (2) |
| C21 | 0.023 (3) | 0.014 (3) | 0.027 (3) | 0.002 (2) | 0.008 (2) | -0.003 (2) |
| C22 | 0.031 (4) | 0.024 (3) | 0.022 (3) | 0.002 (3) | 0.006 (3) | 0.001 (2) |
| C24 | 0.030 (4) | 0.016 (3) | 0.027 (3) | 0.002 (3) | 0.012 (3) | -0.004 (2) |
| C25 | 0.025 (4) | 0.026 (3) | 0.033 (3) | -0.002 (3) | 0.007 (3) | 0.003 (3) |
| C26 | 0.024 (4) | 0.024 (3) | 0.039 (3) | 0.006 (3) | 0.011 (3) | 0.002 (3) |
| C27 | 0.034 (4) | 0.022 (3) | 0.043 (3) | 0.001 (3) | 0.015 (3) | -0.009 (3) |
| C28 | 0.037 (4) | 0.022 (3) | 0.038 (3) | -0.005 (3) | 0.010 (3) | -0.011 (3) |
| C29 | 0.028 (4) | 0.017 (3) | 0.028 (3) | 0.002 (3) | 0.012 (3) | 0.001 (2) |
| C30 | 0.033 (4) | 0.024 (3) | 0.021 (3) | 0.000 (3) | 0.007 (3) | -0.005 (2) |
| C32 | 0.019 (3) | 0.020 (3) | 0.035 (3) | 0.000 (3) | 0.010 (3) | 0.001 (2) |
| N1 | 0.030 (3) | 0.019 (2) | 0.034 (3) | 0.003 (2) | 0.011 (2) | 0.000 (2) |
| N2 | 0.027 (3) | 0.016 (2) | 0.029 (2) | -0.002 (2) | 0.009 (2) | 0.0015 (19) |
| N3 | 0.025 (3) | 0.020 (2) | 0.036 (3) | -0.009 (2) | 0.005 (2) | -0.003 (2) |
| N4 | 0.021 (3) | 0.025 (3) | 0.034 (2) | -0.004 (2) | 0.001 (2) | 0.001 (2) |
| N5 | 0.024 (3) | 0.019 (2) | 0.025 (2) | 0.001 (2) | 0.007 (2) | -0.0008 (19) |
| N6 | 0.026 (3) | 0.021 (2) | 0.028 (2) | 0.001 (2) | 0.011 (2) | 0.003 (2) |
| N7 | 0.023 (3) | 0.028 (3) | 0.035 (3) | -0.005 (2) | 0.005 (2) | -0.006 (2) |
| N8 | 0.032 (3) | 0.042 (3) | 0.057 (3) | -0.006 (3) | 0.017 (3) | -0.016 (3) |
| N9 | 0.020 (3) | 0.019 (2) | 0.031 (2) | -0.002 (2) | 0.007 (2) | -0.0047 (19) |
| N10 | 0.020 (3) | 0.017 (2) | 0.022 (2) | -0.001 (2) | 0.0064 (19) | -0.0004 (18) |
| N11 | 0.022 (3) | 0.026 (3) | 0.023 (2) | -0.010 (2) | 0.000 (2) | 0.0038 (19) |
| N12 | 0.030 (3) | 0.026 (3) | 0.039 (3) | -0.011 (2) | 0.007 (2) | 0.002 (2) |
| N13 | 0.023 (3) | 0.018 (2) | 0.029 (2) | 0.001 (2) | 0.009 (2) | -0.0018 (19) |
| N14 | 0.017 (3) | 0.021 (2) | 0.025 (2) | 0.002 (2) | 0.0043 (19) | -0.0006 (19) |
| N15 | 0.028 (3) | 0.023 (2) | 0.021 (2) | 0.001 (2) | -0.004 (2) | -0.0029 (18) |
| N16 | 0.031 (3) | 0.028 (3) | 0.036 (3) | 0.007 (2) | 0.000 (2) | -0.010 (2) |
| N17 | 0.041 (4) | 0.041 (3) | 0.042 (3) | -0.006 (3) | 0.006 (3) | -0.006 (3) |
| N18 | 0.026 (3) | 0.031 (3) | 0.028 (3) | -0.003 (3) | 0.002 (2) | 0.004 (2) |
| N19 | 0.037 (4) | 0.031 (3) | 0.039 (3) | -0.005 (3) | 0.009 (3) | 0.007 (2) |
| N20 | 0.032 (3) | 0.031 (3) | 0.041 (3) | 0.010 (3) | 0.000 (3) | -0.001 (2) |
| O1 | 0.063 (3) | 0.077 (4) | 0.059 (3) | -0.036 (3) | 0.038 (3) | -0.026 (3) |
| O2 | 0.081 (4) | 0.059 (3) | 0.075 (3) | -0.013 (3) | 0.042 (3) | -0.022 (3) |
| O3 | 0.053 (3) | 0.046 (3) | 0.061 (3) | -0.020 (3) | 0.006 (2) | -0.003 (2) |
| O4 | 0.023 (2) | 0.030 (2) | 0.043 (2) | -0.0044 (19) | 0.0008 (19) | 0.0046 (18) |
| O5 | 0.028 (3) | 0.033 (2) | 0.054 (3) | 0.005 (2) | -0.005 (2) | 0.001 (2) |
| O6 | 0.038 (3) | 0.019 (2) | 0.045 (2) | -0.0062 (19) | 0.012 (2) | -0.0087 (18) |
| O7 | 0.051 (3) | 0.021 (2) | 0.072 (3) | -0.004 (2) | 0.036 (2) | -0.008 (2) |
| O8 | 0.049 (3) | 0.047 (3) | 0.058 (3) | 0.019 (3) | 0.011 (2) | 0.006 (2) |
| O9 | 0.065 (3) | 0.059 (3) | 0.058 (3) | -0.045 (3) | -0.014 (3) | 0.015 (2) |
| O10 | 0.047 (3) | 0.027 (2) | 0.037 (2) | -0.004 (2) | -0.007 (2) | 0.0000 (18) |
| O11 | 0.046 (3) | 0.037 (2) | 0.046 (3) | -0.008 (2) | 0.010 (2) | 0.007 (2) |
| O12 | 0.046 (3) | 0.046 (3) | 0.033 (2) | 0.008 (2) | -0.006 (2) | -0.005 (2) |
| O1W | 0.066 (3) | 0.076 (3) | 0.040 (3) | -0.008 (3) | 0.010 (2) | 0.004 (2) |
| O2W | 0.048 (3) | 0.070 (3) | 0.061 (3) | 0.002 (3) | 0.018 (2) | 0.005 (2) |
| O3W | 0.046 (3) | 0.034 (2) | 0.040 (2) | -0.001 (2) | 0.012 (2) | -0.0074 (19) |
| O4W | 0.046 (3) | 0.053 (3) | 0.055 (3) | -0.010 (2) | -0.008 (2) | 0.004 (2) |

Geometric parameters (Å, °)

| | | | |
|----------|-------------|----------|-----------|
| Zn1—N6 | 2.133 (4) | C21—C22 | 1.463 (7) |
| Zn1—N2 | 2.133 (4) | C22—N10 | 1.283 (6) |
| Zn1—N5 | 2.196 (4) | C22—H22A | 0.9300 |
| Zn1—N1 | 2.211 (4) | C24—N12 | 1.319 (6) |
| Zn1—S1 | 2.4678 (15) | C24—N11 | 1.356 (6) |
| Zn1—S2 | 2.4860 (16) | C25—N13 | 1.331 (6) |
| Zn2—N10 | 2.165 (4) | C25—C26 | 1.389 (7) |
| Zn2—N14 | 2.165 (4) | C25—H25A | 0.9300 |
| Zn2—N13 | 2.178 (4) | C26—C27 | 1.372 (7) |
| Zn2—N9 | 2.203 (4) | C26—H26A | 0.9300 |
| Zn2—S4 | 2.4475 (15) | C27—C28 | 1.393 (7) |
| Zn2—S3 | 2.4733 (15) | C27—H27A | 0.9300 |
| S1—C8 | 1.692 (5) | C28—C29 | 1.380 (7) |
| S2—C16 | 1.689 (6) | C28—H28A | 0.9300 |
| S3—C24 | 1.688 (5) | C29—N13 | 1.356 (6) |
| S4—C32 | 1.690 (5) | C29—C30 | 1.464 (7) |
| C1—N1 | 1.319 (6) | C30—N14 | 1.270 (6) |
| C1—C2 | 1.386 (7) | C30—H30A | 0.9300 |
| C1—H1A | 0.9300 | C32—N16 | 1.304 (6) |
| C2—C3 | 1.385 (7) | C32—N15 | 1.369 (6) |
| C2—H2A | 0.9300 | N2—N3 | 1.361 (5) |
| C3—C4 | 1.372 (7) | N3—H3B | 0.8600 |
| C3—H3A | 0.9300 | N4—H4B | 0.8600 |
| C4—C5 | 1.388 (7) | N4—H4C | 0.8600 |
| C4—H4A | 0.9300 | N6—N7 | 1.367 (5) |
| C5—N1 | 1.349 (6) | N7—H7A | 0.8600 |
| C5—C6 | 1.445 (7) | N8—H8A | 0.8600 |
| C6—N2 | 1.273 (6) | N8—H8B | 0.8600 |
| C6—H6A | 0.9300 | N10—N11 | 1.359 (5) |
| C8—N4 | 1.314 (6) | N11—H11B | 0.8600 |
| C8—N3 | 1.353 (6) | N12—H12B | 0.8600 |
| C9—N5 | 1.322 (6) | N12—H12C | 0.8600 |
| C9—C10 | 1.404 (7) | N14—N15 | 1.348 (5) |
| C9—H9A | 0.9300 | N15—H15A | 0.8600 |
| C10—C11 | 1.375 (7) | N16—H16A | 0.8600 |
| C10—H10A | 0.9300 | N16—H16B | 0.8600 |
| C11—C12 | 1.371 (7) | N17—O1 | 1.233 (6) |
| C11—H11A | 0.9300 | N17—O2 | 1.249 (6) |
| C12—C13 | 1.375 (6) | N17—O3 | 1.250 (6) |
| C12—H12A | 0.9300 | N18—O4 | 1.252 (5) |
| C13—N5 | 1.347 (6) | N18—O5 | 1.256 (5) |
| C13—C14 | 1.452 (7) | N18—O6 | 1.258 (5) |
| C14—N6 | 1.273 (6) | N19—O9 | 1.229 (5) |
| C14—H14A | 0.9300 | N19—O8 | 1.238 (5) |
| C16—N8 | 1.329 (6) | N19—O7 | 1.249 (5) |
| C16—N7 | 1.354 (6) | N20—O12 | 1.236 (5) |

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| C17—N9 | 1.328 (6) | N20—O11 | 1.252 (5) |
| C17—C18 | 1.387 (7) | N20—O10 | 1.281 (5) |
| C17—H17A | 0.9300 | O1W—H1WA | 0.8500 |
| C18—C19 | 1.359 (7) | O1W—H1WB | 0.8501 |
| C18—H18A | 0.9300 | O2W—H2WA | 0.8499 |
| C19—C20 | 1.394 (7) | O2W—H2WB | 0.8500 |
| C19—H19A | 0.9300 | O3W—H3WA | 0.8502 |
| C20—C21 | 1.383 (6) | O3W—H3WB | 0.8500 |
| C20—H20A | 0.9300 | O4W—H4WA | 0.8500 |
| C21—N9 | 1.347 (6) | O4W—H4WB | 0.8502 |
| N6—Zn1—N2 | 167.04 (16) | N10—C22—C21 | 116.6 (5) |
| N6—Zn1—N5 | 74.23 (16) | N10—C22—H22A | 121.7 |
| N2—Zn1—N5 | 97.52 (16) | C21—C22—H22A | 121.7 |
| N6—Zn1—N1 | 95.16 (16) | N12—C24—N11 | 116.2 (5) |
| N2—Zn1—N1 | 74.32 (16) | N12—C24—S3 | 121.4 (4) |
| N5—Zn1—N1 | 88.34 (15) | N11—C24—S3 | 122.4 (4) |
| N6—Zn1—S1 | 110.45 (11) | N13—C25—C26 | 123.0 (5) |
| N2—Zn1—S1 | 79.34 (12) | N13—C25—H25A | 118.5 |
| N5—Zn1—S1 | 91.99 (11) | C26—C25—H25A | 118.5 |
| N1—Zn1—S1 | 153.47 (13) | C27—C26—C25 | 119.1 (5) |
| N6—Zn1—S2 | 78.96 (12) | C27—C26—H26A | 120.5 |
| N2—Zn1—S2 | 108.58 (11) | C25—C26—H26A | 120.5 |
| N5—Zn1—S2 | 153.14 (12) | C26—C27—C28 | 118.9 (5) |
| N1—Zn1—S2 | 92.62 (11) | C26—C27—H27A | 120.6 |
| S1—Zn1—S2 | 98.79 (5) | C28—C27—H27A | 120.6 |
| N10—Zn2—N14 | 169.14 (15) | C29—C28—C27 | 118.6 (5) |
| N10—Zn2—N13 | 97.54 (15) | C29—C28—H28A | 120.7 |
| N14—Zn2—N13 | 73.92 (16) | C27—C28—H28A | 120.7 |
| N10—Zn2—N9 | 73.97 (15) | N13—C29—C28 | 122.7 (5) |
| N14—Zn2—N9 | 99.09 (15) | N13—C29—C30 | 115.1 (4) |
| N13—Zn2—N9 | 90.90 (15) | C28—C29—C30 | 122.1 (5) |
| N10—Zn2—S4 | 109.51 (11) | N14—C30—C29 | 116.6 (5) |
| N14—Zn2—S4 | 78.71 (11) | N14—C30—H30A | 121.7 |
| N13—Zn2—S4 | 152.61 (12) | C29—C30—H30A | 121.7 |
| N9—Zn2—S4 | 92.42 (11) | N16—C32—N15 | 116.1 (5) |
| N10—Zn2—S3 | 78.04 (11) | N16—C32—S4 | 121.2 (4) |
| N14—Zn2—S3 | 108.77 (11) | N15—C32—S4 | 122.7 (4) |
| N13—Zn2—S3 | 93.90 (11) | C1—N1—C5 | 118.7 (5) |
| N9—Zn2—S3 | 151.99 (12) | C1—N1—Zn1 | 127.7 (4) |
| S4—Zn2—S3 | 95.80 (5) | C5—N1—Zn1 | 113.5 (3) |
| C8—S1—Zn1 | 97.99 (18) | C6—N2—N3 | 121.0 (4) |
| C16—S2—Zn1 | 97.7 (2) | C6—N2—Zn1 | 118.6 (4) |
| C24—S3—Zn2 | 98.53 (19) | N3—N2—Zn1 | 120.4 (3) |
| C32—S4—Zn2 | 99.06 (19) | C8—N3—N2 | 119.7 (4) |
| N1—C1—C2 | 122.7 (6) | C8—N3—H3B | 120.2 |
| N1—C1—H1A | 118.6 | N2—N3—H3B | 120.2 |
| C2—C1—H1A | 118.6 | C8—N4—H4B | 120.0 |
| C3—C2—C1 | 119.0 (6) | C8—N4—H4C | 120.0 |
| C3—C2—H2A | 120.5 | H4B—N4—H4C | 120.0 |

| | | | |
|--------------|-----------|---------------|-----------|
| C1—C2—H2A | 120.5 | C9—N5—C13 | 118.6 (4) |
| C4—C3—C2 | 118.5 (5) | C9—N5—Zn1 | 126.8 (3) |
| C4—C3—H3A | 120.7 | C13—N5—Zn1 | 114.6 (3) |
| C2—C3—H3A | 120.7 | C14—N6—N7 | 120.2 (5) |
| C3—C4—C5 | 119.4 (6) | C14—N6—Zn1 | 118.6 (4) |
| C3—C4—H4A | 120.3 | N7—N6—Zn1 | 121.1 (3) |
| C5—C4—H4A | 120.3 | C16—N7—N6 | 118.8 (4) |
| N1—C5—C4 | 121.7 (5) | C16—N7—H7A | 120.6 |
| N1—C5—C6 | 116.2 (5) | N6—N7—H7A | 120.6 |
| C4—C5—C6 | 122.1 (5) | C16—N8—H8A | 120.0 |
| N2—C6—C5 | 117.2 (5) | C16—N8—H8B | 120.0 |
| N2—C6—H6A | 121.4 | H8A—N8—H8B | 120.0 |
| C5—C6—H6A | 121.4 | C17—N9—C21 | 118.1 (4) |
| N4—C8—N3 | 115.3 (5) | C17—N9—Zn2 | 127.2 (3) |
| N4—C8—S1 | 122.1 (4) | C21—N9—Zn2 | 114.7 (3) |
| N3—C8—S1 | 122.6 (4) | C22—N10—N11 | 120.9 (4) |
| N5—C9—C10 | 121.5 (5) | C22—N10—Zn2 | 117.9 (4) |
| N5—C9—H9A | 119.3 | N11—N10—Zn2 | 121.1 (3) |
| C10—C9—H9A | 119.3 | C24—N11—N10 | 118.5 (4) |
| C11—C10—C9 | 119.1 (5) | C24—N11—H11B | 120.7 |
| C11—C10—H10A | 120.5 | N10—N11—H11B | 120.7 |
| C9—C10—H10A | 120.5 | C24—N12—H12B | 120.0 |
| C12—C11—C10 | 119.3 (5) | C24—N12—H12C | 120.0 |
| C12—C11—H11A | 120.4 | H12B—N12—H12C | 120.0 |
| C10—C11—H11A | 120.4 | C25—N13—C29 | 117.7 (4) |
| C11—C12—C13 | 118.5 (5) | C25—N13—Zn2 | 126.7 (3) |
| C11—C12—H12A | 120.7 | C29—N13—Zn2 | 115.6 (3) |
| C13—C12—H12A | 120.7 | C30—N14—N15 | 120.2 (4) |
| N5—C13—C12 | 123.0 (5) | C30—N14—Zn2 | 118.6 (4) |
| N5—C13—C14 | 115.3 (4) | N15—N14—Zn2 | 121.1 (3) |
| C12—C13—C14 | 121.7 (5) | N14—N15—C32 | 118.4 (4) |
| N6—C14—C13 | 117.2 (5) | N14—N15—H15A | 120.8 |
| N6—C14—H14A | 121.4 | C32—N15—H15A | 120.8 |
| C13—C14—H14A | 121.4 | C32—N16—H16A | 120.0 |
| N8—C16—N7 | 115.1 (5) | C32—N16—H16B | 120.0 |
| N8—C16—S2 | 121.5 (4) | H16A—N16—H16B | 120.0 |
| N7—C16—S2 | 123.3 (4) | O1—N17—O2 | 120.8 (5) |
| N9—C17—C18 | 123.4 (5) | O1—N17—O3 | 120.7 (5) |
| N9—C17—H17A | 118.3 | O2—N17—O3 | 118.5 (6) |
| C18—C17—H17A | 118.3 | O4—N18—O5 | 120.5 (4) |
| C19—C18—C17 | 118.2 (5) | O4—N18—O6 | 120.0 (5) |
| C19—C18—H18A | 120.9 | O5—N18—O6 | 119.5 (5) |
| C17—C18—H18A | 120.9 | O9—N19—O8 | 120.5 (5) |
| C18—C19—C20 | 119.8 (5) | O9—N19—O7 | 120.2 (6) |
| C18—C19—H19A | 120.1 | O8—N19—O7 | 119.3 (5) |
| C20—C19—H19A | 120.1 | O12—N20—O11 | 121.5 (5) |
| C21—C20—C19 | 118.3 (5) | O12—N20—O10 | 120.1 (5) |
| C21—C20—H20A | 120.9 | O11—N20—O10 | 118.4 (5) |
| C19—C20—H20A | 120.9 | H1WA—O1W—H1WB | 112.2 |

supplementary materials

| | | | |
|-------------|-----------|---------------|-------|
| N9—C21—C20 | 122.1 (5) | H2WA—O2W—H2WB | 127.8 |
| N9—C21—C22 | 115.9 (4) | H3WA—O3W—H3WB | 100.8 |
| C20—C21—C22 | 121.9 (5) | H4WA—O4W—H4WB | 108.9 |

Hydrogen-bond geometry (\AA , $^\circ$)

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--------------------------------------|-------|-------------|-------------|---------------|
| N3—H3B \cdots O7 | 0.86 | 2.19 | 2.966 (6) | 150 |
| N4—H4B \cdots O7 | 0.86 | 2.17 | 2.957 (6) | 153 |
| N4—H4C \cdots O3W | 0.86 | 2.03 | 2.840 (5) | 157 |
| N7—H7A \cdots O1 | 0.86 | 1.96 | 2.813 (6) | 172 |
| N8—H8A \cdots O2 | 0.86 | 2.08 | 2.906 (6) | 160 |
| N8—H8B \cdots O6 ⁱ | 0.86 | 2.04 | 2.870 (6) | 163 |
| N11—H11B \cdots O10 | 0.86 | 1.87 | 2.710 (5) | 167 |
| N12—H12B \cdots O11 | 0.86 | 2.02 | 2.876 (6) | 173 |
| N12—H12C \cdots O8 ⁱⁱ | 0.86 | 2.27 | 3.114 (6) | 168 |
| N15—H15A \cdots O4W ⁱⁱⁱ | 0.86 | 1.87 | 2.733 (5) | 178 |
| N16—H16A \cdots O6 ⁱⁱⁱ | 0.86 | 2.11 | 2.909 (6) | 155 |
| N16—H16B \cdots O4 ^{iv} | 0.86 | 2.02 | 2.873 (5) | 173 |
| O1W—H1WA \cdots O4 ^v | 0.85 | 2.32 | 3.042 (5) | 144 |
| O1W—H1WB \cdots O3 | 0.85 | 2.39 | 3.041 (6) | 134 |
| O2W—H2WA \cdots O5 ^{iv} | 0.85 | 2.33 | 2.927 (6) | 128 |
| O2W—H2WB \cdots O3 | 0.85 | 2.47 | 3.141 (6) | 136 |
| O3W—H3WA \cdots O10 ^{vi} | 0.85 | 2.22 | 2.800 (5) | 126 |
| O3W—H3WB \cdots O9 ^{vii} | 0.85 | 2.24 | 2.978 (6) | 145 |
| O4W—H4WA \cdots O2W | 0.85 | 2.09 | 2.746 (6) | 133 |
| O4W—H4WB \cdots O6 | 0.85 | 2.55 | 2.958 (5) | 111 |

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $x, -y+1/2, z+1/2$; (iii) $x, -y+3/2, z+1/2$; (iv) $-x+1, y-1/2, -z+3/2$; (v) $x, -y+3/2, z-1/2$; (vi) $-x, y-1/2, -z+1/2$; (vii) $-x, y+1/2, -z+1/2$.

Fig. 1

